metal-organic papers

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Key indicators

Single-crystal X-ray study T = 173 K Mean σ (S–C) = 0.002 Å R factor = 0.018 wR factor = 0.041 Data-to-parameter ratio = 32.7

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

cis-trans-Tetrachlorobis(dimethyl sulfide-kS)tin

The molecule of the title compound, $[SnCl_4(C_2H_6S)_2]$, is located on a centre of inversion with a half-molecule in the asymmetric unit. The Sn atom shows a quadratic bipyramidal coordination. Received 17 August 2005 Accepted 18 August 2005 Online 27 August 2005

Comment

Recently, we have reported on the synthesis and structure of the adduct of Me₃SnCl with Me₃SnOH and H₂O. This adduct represents an intermediate in Me₃SnCl hydrolysis. The structure of Me₃SnCl·Me₃SnOH·H₂O features an array of Me₃Sn units connected alternately by bridging Cl and OH ligands (Lerner *et al.*, 2005). We report here the X-ray crystal structure analysis of the adduct [SnCl₄]·[CH₃SCH₃]₂, (I). The synthesis of (I) was achieved by treatment of SnCl₄ with 2 equivalents of CH₃SCH₃, as indicated in the scheme below.





A perspective view of (I) is shown in Fig. 1 and a packing diagram is shown in Fig. 2. Bond lengths and angles can be regarded as normal (Cambridge Structural Database, Version 1.7 plus three updates; Allen, 2002). The coordination mode of the Sn atom is almost perfect quadratic bipyramidal. The two dimethyl sulfide residues occupy the apical positions and the Cl ligands are located in the quadratic plane. One Cl ligand almost bisects the CH_3-S-CH_3 angle (Table 1).

It is remarkable that *cis-trans*-tetrabromo-bis(dimethylthio)tin, (II) (Bricklebank *et al.*, 1994), is not isostructural with the title compound. Nevertheless, the cell parameters show some similarities: after transforming the cell of (II) from $P2_1/c$ into $P2_1/n$, the cell parameters are: a = 7.617 Å, b = 12.382 Å, c = 23.209 Å and $\beta = 102.98^{\circ}$. Unfortunately, no coordinates are available for (II). Therefore, the two structures cannot be compared.

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Experimental

 $SnCl_4$ (0.71 ml) was added with stirring at ambient temperature to CH_3SCH_3 (25 ml). Colourless crystals of the title compound were grown by storing this solution at room temperature for several days.

 $D_{\rm r} = 2.021 {\rm Mg m}^{-3}$

Cell parameters from 15669

1798 independent reflections

1673 reflections with $I > 2\sigma(I)$

Mo $K\alpha$ radiation

reflections

 $\theta = 3.5 - 29.9^{\circ}$ $\mu = 3.14 \text{ mm}^{-1}$

T = 173 (2) K

 $R_{\rm int} = 0.040$

 $\theta_{\rm max} = 29.7^{\circ}$ $h = -9 \rightarrow 9$

 $k = -17 \rightarrow 17$

 $l = -11 \rightarrow 11$

Block, colourless $0.10 \times 0.07 \times 0.04 \text{ mm}$

Crystal data

 $[SnCl_4(C_2H_6S)_2]$ $M_r = 384.75$ Monoclinic, $P2_1/n$ a = 6.6711 (5) Å b = 12.2317 (8) Å c = 8.1051 (6) Å $\beta = 107.068$ (6)° V = 632.24 (8) Å³ Z = 2

Data collection

Stoe IPDS-II two-circle diffractometer ω scans Absorption correction: multi-scan (*MULABS*; Spek, 2003; Blessing, 1995) $T_{\min} = 0.744, T_{\max} = 0.885$ 15669 measured reflections

Refinement

Refinement on F^2	$w = 1/[\sigma^2(F_0^2) + (0.0222P)^2]$
$R[F^2 > 2\sigma(F^2)] = 0.018$	+ 0.192P]
$wR(F^2) = 0.042$	where $P = (F_0^2 + 2F_c^2)/3$
S = 1.13	$(\Delta/\sigma)_{\rm max} = 0.001$
1798 reflections	$\Delta \rho_{\rm max} = 0.45 \ {\rm e} \ {\rm \AA}^{-3}$
55 parameters	$\Delta \rho_{\rm min} = -0.48 \ {\rm e} \ {\rm \AA}^{-3}$
H-atom parameters constrained	Extinction correction: SHELXL97
-	Extinction coefficient: 0.0191 (10)

Table 1			
Selected geometric parameters	(Å,	°).	

Sn1-Cl2 Sn1-Cl1	2.4147 (4) 2.4188 (5)	Sn1-S1	2.6208 (4)
$\begin{array}{c} Cl2 - Sn1 - Cl1^{i} \\ Cl2 - Sn1 - Cl1 \\ Cl2 - Sn1 - S1 \end{array}$	89.669 (17) 90.332 (17) 91.174 (14)	Cl1-Sn1-S1 C1-S1-C2	89.120 (14) 99.84 (9)
Cl1 ⁱ -Sn1-S1-C1	56.15 (7)	$Cl1^i - Sn1 - S1 - C2$	-47.87 (7)

Symmetry code: (i) -x + 1, -y + 1, -z + 1.

H atoms were located in a difference map and refined with fixed individual displacement parameters $[U_{\rm iso}({\rm H}) = 1.5U_{\rm eq}({\rm C})]$, using a riding model, with C-H = 0.98 Å. The methyl groups were allowed to rotate but not to tip.

Data collection: X-AREA (Stoe & Cie, 2001); cell refinement: X-AREA; data reduction: X-AREA; program(s) used to solve structure: SHELXS97 (Sheldrick, 1990); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: XP in SHELXTL-Plus (Sheldrick, 1991); software used to prepare material for publication: SHELXL97.

References

Allen, F. H. (2002). Acta Cryst. B58, 380–388. Blessing, R. H. (1995). Acta Cryst. A51, 33–38.



Figure 1

Perspective view of the title compound, with the atom numbering; displacement ellipsoids are drawn at the 50% probability level for non-H atoms. [Symmetry code: (i) -x + 1, -y + 1, -z + 1.]



Figure 2

The packing of the title compound, viewed along the *a* axis.

- Bricklebank, N., Godfrey, S. M., McAuliffe, C. A. & Pritchard, R. G. (1994). *Chem. Commun.* pp. 695–696.
- Lerner, H.-W., Haghiri Ilkechi, A., Bolte, M. & Wagner, M. (2005). Z. Naturforsch. Teil B, 60, 413-415.
- Sheldrick, G. M. (1990). Acta Cryst. A46, 467-473.
- Sheldrick, G. M. (1991). *SHELXTL-Plus*. Release 4.1. Siemens Analytical X-ray Instruments Inc., Madison, Wisconsin, USA.
- Sheldrick, G. M. (1997). SHELXL97. University of Göttingen, Germany. Spek, A. L. (2003). J. Appl. Cryst. 36, 7–13.
- Stoe & Cie (2001). X-AREA. Stoe & Cie, Darmstadt, Germany.